Two New Ceramides from the Marine Sponge Ircinia fasciculata

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A new 4-sulfated ceramide, ircisulfamide (= N-[($1S^*$, $2S^*$, $3R^*$)-2-hydroxy-1-(hydroxymethyl)-3-(sulfooxy)-heptadecyl]hexadecanamide; **1**), and a new glycosphingolipid, ircicerebroside (= $(2R^*)$ -N-{($1S^*$, $2R^*$,3E,7E)-1-[(β -D-glucopyranosyloxy)methyl]-2-hydroxy-8-methylheptadeca-3,7-dienyl}-2-hydroxyeicosanamide; **2**), were isolated from the aqueous EtOH extract of the marine sponge *Ircinia fasciculata* (Pallas). The structures of the new compounds were elucidated on the basis of spectroscopic analysis and by means of chemical methods.

Introduction. – Different sphingolipids such as ceramide, sphingosine, and sphingosine-1-phosphate have been isolated from marine vertebrates, invertebrates, algae, and fungi *etc.* [1–3]. These compounds have received increasing attention in the last years as some of their derivatives act as endogenous cell-function modulators and secondary messengers [4]. Sphingolipids have been shown to be potent and specific inhibitors of protein kinase C [5], Na- and K-ATPase, and calmodulin kinase, and to activate phospholipase C, phospholipase D, casein kinase II, tyrosine kinase, and DG kinase [6]. In addition, sphingolipids can also induce Ca²⁺ release [7], proliferate cells [8], protect human keratinocytes from apoptosis [9], suppress colon carcinogenesis in rats [10], and enhance apoptosis of radiation-resistant prostate cancer cells [11].

In continuation of our search for potent bioactive secondary metabolites from the marine invertebrates [12–14], a new 4-sulfated ceramide, ircisulfamide (1), and a new glycosphingolipid, ircicerebroside (2), were isolated from the aqueous EtOH extract of the marine sponge *Ircinia fasciculata*. The structures of compounds 1 and 2 were elucidated on the basis of spectroscopic analysis and chemical methods.

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Results and Discussion. – Compound **1** was isolated as a colorless, amorphous, optically active powder, with $[a]_D^{20} = +16.7$ (c = 0.24, CHCl₃). From HR-FAB-MS experiments, the molecular formula $C_{34}H_{69}NO_7S$ was deduced, corresponding to three degrees of unsaturation. The IR absorption at 1234 cm⁻¹ was attributed to a sulfate functionality, as further confirmed by elemental analysis. The IR absorption band at 3383 cm⁻¹ indicated OH and NH groups, and the typical absorptions at 1645 and 1545 cm⁻¹ suggested that **1** is a secondary amide, which was supported by NMR experiments ($\delta(H)$ 5.97 (d, J = 6.0 Hz, NH); $\delta(C)$ 51.9 (N-C); $\delta(C)$ 174.0 (C=O)) (see *Table 1*). The ¹H-NMR spectrum of **1** revealed the presence of two Me groups at $\delta(H)$ 0.88 (2t), an N-CH function at $\delta(H)$ 4.38, two O-CH groups at $\delta(H)$ 3.97 and 3.70, an O-CH₂ moiety at $\delta(H)$ 4.18 and 3.51, and a series of overlapped CH₂ resonances at $\delta(H)$ 1.20-1.34 (*Table 1*). The ¹³C-NMR spectrum revealed the presence of two oxygenated C-atoms at $\delta(C)$ 70.4 (t) and 75.2 (t). Besides, a low-field signal at $\delta(C)$ 85.2 (t) suggested the presence of a sulfate group. All the spectroscopic data confirmed that **1** was a ceramide derivative.

	¹³ C	$^{1}\mathrm{H}$	¹ H, ¹ H-COSY
$H_a-C(1)$	70.4 (t)	4.18 (dd, J = 9.0, 7.0)	H-C(2)
$H_b-C(1)$	_	3.51 (dd, J = 9.0, 7.0)	_
H-C(2)	51.9 (d)	4.38 (m)	H-C(1), H-C(3), NH
H-C(3)	75.2 (d)	3.97 (dd, J = 6.0, 4.0)	H-C(2), H-C(4)
H-C(4)	85.2 (d)	3.70 (dd, J = 11.5, 6.0)	H-C(3), H-C(5)
$CH_2(5)$	33.4 (t)	1.56 (m)	H-C(4)
$CH_2(6-16)$	29.3 - 29.7(t)	1.20 - 1.34	_
$CH_2(17)$	22.7 (t)	_	_
Me(18)	14.1 (q)	0.88(t, J=7.0)	_
NH	_	5.97 (d, J = 6.0)	H-C(2)
C(1')	174.0 (s)	=	_
$CH_2(2')$	36.7(t)	2.22(t, J=7.5)	H-C(3')
$CH_2(3')$	25.8(t)	1.63 (m)	H-C(2')
$CH_2(4'-14')$	29.3-29.7(t)	1.20 - 1.34	_
CH ₂ (15')	22.7 (t)	_	-
Me(16')	14.1 (q)	0.88(t, J=7.0)	_

Table 1. NMR Data for Ircisulfamide (1). In CDCl₃; δ in ppm, J in Hz. Arbitrary atom numbering.

 1 H, 1 H-COSY Correlations revealed two different networks, as shown in the *Figure*. In the HMQC spectrum, the signal at $\delta(H)$ 3.70 (m, H-C(4)) gave a cross-peak with the signal at $\delta(C)$ 85.2, indicating that the OSO₃H group was linked at $C(4)^{1}$). The HMBC correlations between C(2)/NH and C(1')/NH not only confirmed the position of the NH group, but also connected the two partial structures *via* an amide bond.

Methanolysis [15] of **1** afforded methyl palmitate, which was identified by GC/MS analysis (m/z 270 (95%)), implying the presence of a hexadecanoyl group. As the formula was $C_{34}H_{69}NO_7S$, the structure of **1** was, assigned as N-[2-hydroxy-1-(hydroxymethyl)-3-(sulfooxy)heptadecyl]hexadecanamide, and named *ircisulfamide*.

¹⁾ Arbitrary atom numbering.

Figure. ¹H, ¹H-COSY and key HMBC correlations for 1 and 2

By considering biogenetic [16] and steric factors, and from the NMR chemical shifts of H-C(2) and those of C(2) and C(3), the relative configuration of $\bf 1$ could be derived [3]. Solvolysis [17] of $\bf 1$ afforded the desulfated ceramide $\bf 1a$, which has been reported before [15]. Its resonances at $\delta(C)$ 62.0 (C(1)) 52.8 (C(2)), 76.7 (C(3)), and 72.3 (C(4)) were very similar to the ones reported (61.9, 52.7, 76.5, and 71.9, resp.) [15], indicating the same relative configuration, *i.e.*, (2S,3S,4R)¹). This was further confirmed by comparison of the optical rotations, with an $[\alpha]_D^{20}$ value of + 19.3 (c=0.01, CHCl₃) for $\bf 1a$.

Whereas 1-sulfated ceramides [17-19] and 4-sulfated sphingosine [20] have been reported, ircisulfamide (1) is, to the best of our knowledge, the first 4-sulfated ceramide.

Compound **2** was obtained as a colorless, amorphous powder, with an $[\alpha]_D^{20}$ value of +9.1 (c=0.02, MeOH). HR-FAB-MS furnished the molecular formula $C_{45}H_{85}NO_9$, with four degrees of unsaturation. The IR absorption band at 3378 cm⁻¹ indicated the presence of OH and amide groups. The typical IR absorptions at 1643 and 1536 cm⁻¹ again suggested a secondary amide, as supported by the presence of NMR signals at $\delta(H)$ 7.73 (d, J = 8.0 Hz, NH), $\delta(C)$ 54.5 (C-N), and $\delta(C)$ 177.1 (C=O) (see *Table 2*). In the ¹H-NMR spectrum, one olefinic signal at $\delta(H)$ 5.14 (m, H-C(8)), assignable to a trisubstituted C=C bond, and two olefinic signals ($\delta(H)$ 5.47 (dd, J=15.5, 7.5 Hz, 1 H); 5.73 (dt, J=15.5, 6.0 Hz, 1 H)), attributable to a disubstituted C=C bond, were observed. The ¹H-NMR spectrum also showed the presence of three Me groups ($\delta(H)$ 1.59 (s, Me(19)); 0.90 (s, Me(18), Me(20'))). As for **1**, two regions corresponding to aliphatic CH₂ groups at $\delta(H)$ 1.20–1.35 were observed. However, in the case of **2**, an additional glucopyranosyl (Glc) moiety could be distinguished.

Methanolysis of **2** afforded methyl 2-hydroxyicosanoate, which was identified by GC/MS analysis (m/z 342 (93%)), together with methyl glucopyranoside [21]. The optical rotation of the latter, $[\alpha]_D^{20} = +76.2$ (c = 0.02, MeOH), was close to that of an authentic D-configured sample. The anomeric signal at $\delta(C)$ 104.6 (d) confirmed that **2** was a monoglycoside, and the signals at $\delta(C)$ 104.6, 74.9, 77.8, 71.5, 77.9, and 62.6 suggested a β -D-glucopyranoside. The coupling constant between the anomeric H-C(1") resonance at $\delta(H)$ 4.26 (d, J = 9.0 Hz) and the vicinal H-C(2") signal at $\delta(H)$ 3.19 (d, J = 9.5 Hz) further supported β -configuration.

 1 H, 1 H-COSY Experiments revealed three distinct networks in **2** (see the *Figure*), and their linkages were resolved by means of HMBC analysis. The large vicinal coupling constant J(4,5) (15.5 Hz) clearly indicated an (E)-configured C=C bond in 4-position. The second unsaturation at C(8) was also assigned (E)-configuration on the

Table 2. NMR Data for Ircicerebroside (2). In CD₃OD; δ in ppm, J in Hz. Arbitrary atom numbering.

$\begin{array}{cccccccccccccccccccccccccccccccccccc$			() - 3-) - FF) -	8
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		¹³ C	¹ H	¹H,¹H-COSY
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H_a -C(1)	69.7 (t)	$4.10 \; (dd, J = 10.0, 5.0)$	$H_b-C(1), H-C(2)$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$H_b-C(1)$	-	3.69 (dd, J = 10.0, 3.5)	$H_a-C(1)$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(2)	54.5 (d)	3.99 (m, 1 H)	H-C(1), H-C(3), NH
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(3)	72.8(d)	4.14 (dd, J = 7.5, 5.0)	H-C(2), H-C(4)
$\begin{array}{c} \mathrm{CH}_2(6) & 33.8 \ (t) & 2.06 \ (m) & \mathrm{H-C(5)}, \mathrm{H-C(7)} \\ \mathrm{CH}_2(7) & 35.4 \ (t) & 2.04 \ (m) & \mathrm{H-C(6)}, \mathrm{H-C(8)} \\ \mathrm{H-C(8)} & 124.8 \ (d) & 5.14 \ (t, J=6.5) & \mathrm{H-C(7)}, \mathrm{H-C(19)} \\ \mathrm{C(9)} & 136.6 \ (s) & - & \mathrm{H-C(8)}, \mathrm{H-C(10)} \\ \mathrm{CH}_2(10) & 40.8 \ (t) & 1.98 \ (t, J=7.5) & - \\ \mathrm{CH}_2(11-16) & 30.4-30.8 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(17) & 23.7 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(17) & 23.7 \ (t) & 1.20-1.35 & - \\ \mathrm{Me}(18) & 14.5 \ (q) & 0.90 \ (t, J=7.0) & - \\ \mathrm{Me}(19) & 16.2 \ (q) & 1.59 \ (s) & - \\ \mathrm{NH} & - & 7.73 \ (d, J=8.0) & - \\ \mathrm{C(1')} & 177.1 \ (s) & - & - \\ \mathrm{H-C(2')} & 73.0 \ (d) & 3.97 \ (m) & - \\ \mathrm{H_a-C(3')} & 35.8 \ (t) & 1.55 \ (m) & \mathrm{H-C(2')} \\ \mathrm{H_b-C(3')} & - & 1.69 \ (m) & \mathrm{H-C(2')}, \mathrm{H_a-C(3')} \\ \mathrm{CH}_2(4') & 26.2 \ (t) & 1.38 \ (m) & - \\ \mathrm{CH}_2(5'-18') & 30.4-30.8 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(19') & 23.7 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(19') & 23.7 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(19') & 23.7 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(1'') & 104.6 \ (d) & 4.26 \ (d, J=9.0) & - \\ \mathrm{H-C(2'')} & 74.9 \ (d) & 3.19 \ (dd, J=9.5, 8.0) & \mathrm{H-C(2'')}, \mathrm{H-C(3'')} \\ \mathrm{H-C(3'')} & 77.8 \ (d) & 3.36 \ (dd, J=9.5, 8.0) & \mathrm{H-C(2'')}, \mathrm{H-C(1'')}, \mathrm{H-C(3'')} \\ \mathrm{H-C(4'')} & 71.5 \ (d) & 3.26 \ (m) & \mathrm{H-C(2'')}, \mathrm{H-C(4'')} \\ \mathrm{H-C(5'')} & 77.9 \ (d) & 3.31 \ (m) & \mathrm{H-C(4'')}, \mathrm{H-C(6'')} \end{array}$	H-C(4)	131.1 (d)	5.47 (dd, J = 15.5, 7.5)	H-C(3), H-C(5), H-C(6)
$\begin{array}{c} \mathrm{CH}_2(7) & 35.4 \ (t) & 2.04 \ (m) & \mathrm{H-C}(6) \ , \mathrm{H-C}(8) \\ \mathrm{H-C}(8) & 124.8 \ (d) & 5.14 \ (t, J=6.5) & \mathrm{H-C}(7) \ , \mathrm{H-C}(19) \\ \mathrm{C}(9) & 136.6 \ (s) & - & \mathrm{H-C}(8) \ , \mathrm{H-C}(10) \\ \mathrm{CH}_2(10) & 40.8 \ (t) & 1.98 \ (t, J=7.5) & - \\ \mathrm{CH}_2(11-16) & 30.4-30.8 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(17) & 23.7 \ (t) & 1.20-1.35 & - \\ \mathrm{Me}(18) & 14.5 \ (q) & 0.90 \ (t, J=7.0) & - \\ \mathrm{Me}(19) & 16.2 \ (q) & 1.59 \ (s) & - \\ \mathrm{NH} & - & 7.73 \ (d, J=8.0) & - \\ \mathrm{C}(1') & 177.1 \ (s) & - & - \\ \mathrm{H-C}(2') & 73.0 \ (d) & 3.97 \ (m) & - \\ \mathrm{H_3-C}(3') & 35.8 \ (t) & 1.55 \ (m) & \mathrm{H-C}(2') \\ \mathrm{H_b-C}(3') & - & 1.69 \ (m) & \mathrm{H-C}(2') \ , \mathrm{H_a-C}(3') \\ \mathrm{CH}_2(4') & 26.2 \ (t) & 1.38 \ (m) & - \\ \mathrm{CH}_2(5'-18') & 30.4-30.8 \ (t) & 1.20-1.35 & - \\ \mathrm{CH}_2(19') & 23.7 \ (t) & 1.20-1.35 & - \\ \mathrm{Me}(20') & 14.4 \ (q) & 0.90 \ (t, J=7.0) & \mathrm{H-C}(2') \ , \mathrm{H_b-C}(3') \\ \mathrm{H-C}(1'') & 104.6 \ (d) & 4.26 \ (d, J=9.0) & - \\ \mathrm{H-C}(2'') & 74.9 \ (d) & 3.19 \ (dd, J=9.5, 8.0) & \mathrm{H-C}(2'') \\ \mathrm{H-C}(3'') & 77.8 \ (d) & 3.26 \ (m) & \mathrm{H-C}(2'') \ , \mathrm{H-C}(4'') \\ \mathrm{H-C}(5'') & 77.9 \ (d) & 3.31 \ (m) & \mathrm{H-C}(2'') \ , \mathrm{H-C}(5'') \\ \mathrm{H_a-C}(6'') & 62.6 \ (t) & 3.87 \ (dd, J=10.5, 2.0) & \mathrm{H-C}(4'') \ , \mathrm{H-C}(6'') \end{array}$	H-C(5)	134.6 (d)	5.73 (dt, J = 15.5, 6.0)	H-C(4), H-C(6)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$CH_2(6)$	33.8 (t)	2.06(m)	H-C(5), H-C(7)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$CH_2(7)$	35.4 (t)	2.04(m)	H-C(6), H-C(8)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(8)	124.8 (d)	5.14 (t, J = 6.5)	H-C(7), H-C(19)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C(9)	136.6 (s)	_	H-C(8), H-C(10)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$CH_2(10)$	40.8(t)	1.98 (t, J = 7.5)	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$CH_2(11-16)$	30.4 - 30.8(t)	1.20 - 1.35	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$CH_2(17)$	23.7(t)	1.20 - 1.35	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Me(18)	14.5 (q)	0.90 (t, J = 7.0)	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Me(19)	16.2 (q)	1.59(s)	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	NH	-	7.73 (d, J = 8.0)	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C(1')	177.1 (s)	_	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(2')	73.0(d)	3.97 (m)	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$H_a-C(3')$	35.8 (t)	1.55(m)	H-C(2')
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$H_b - C(3')$	-	1.69 (m)	$H-C(2'), H_a-C(3')$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$CH_2(4')$	26.2 (t)	1.38 (m)	_
$\begin{array}{llllllllllllllllllllllllllllllllllll$	$CH_2(5'-18')$	30.4 - 30.8(t)	1.20 - 1.35	_
$\begin{array}{llllllllllllllllllllllllllllllllllll$	$CH_2(19')$	23.7(t)	1.20 - 1.35	_
$\begin{array}{llllllllllllllllllllllllllllllllllll$	Me(20')	14.4 (q)	0.90 (t, J = 7.0)	$H-C(2'), H_b-C(3')$
$\begin{array}{llllllllllllllllllllllllllllllllllll$	H-C(1'')	104.6 (d)	4.26 (d, J = 9.0)	_
H-C(4'') 71.5 (d) 3.26 (m) $H-C(2'')$, $H-C(4'')$ $H-C(5'')$ 77.9 (d) 3.31 (m) $H-C(5'')$ $H-C(5'')$ 62.6 (t) 3.87 (dd, $J=10.5, 2.0$) $H-C(4'')$, $H-C(6'')$	H-C(2'')	74.9(d)	3.19 (dd, J = 9.5, 8.0)	H-C(2'')
$H-C(5'')$ 77.9 (d) 3.31 (m) $H-C(3'')$, $H-C(5'')$ $H_a-C(6'')$ 62.6 (t) 3.87 (dd, $J=10.5, 2.0$) $H-C(4'')$, $H-C(6'')$	H-C(3'')	77.8(d)	3.36 (dd, J = 9.5, 8.0)	H-C(1''), H-C(3'')
$H_a - C(6'')$ 62.6 (t) 3.87 (dd, $J = 10.5, 2.0$) $H - C(4''), H - C(6'')$	H-C(4'')	71.5 (d)	3.26 (m)	H-C(2''), H-C(4'')
	H-C(5'')	77.9(d)	3.31 (m)	H-C(3''), H-C(5'')
$H_b-C(6'')$ - 3.67 (dd, $J=10.5, 3.5$) $H-C(5''), H_b-C(6'')$	$H_a - C(6'')$	62.6 (t)	3.87 (dd, J = 10.5, 2.0)	
	$H_b-C(6'')$	-	3.67 (dd, J = 10.5, 3.5)	$H-C(5''), H_b-C(6'')$

basis of the upfield-shifted ¹³C-NMR signal for Me(19) [22]. By comparing the NMR chemical shifts of H-C(2) and C(1) to C(3), C(1'), and C(2') with those of known glucosphingolipids [3][21], the relative configuration of **2** was determined as $(2S,3R,2'R)^1$). Compound **2** was, thus, assigned the structure $(2R^*)-N-\{(1S^*,2R^*,3E,7E)-1-[(\beta-D-glucopyranosyloxy)methyl]-2-hydroxy-8-methylheptadeca-3,7-dienyl}-2-hydroxyeicosanamide, and named$ *ircicerebroside*.

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Experimental Part

General. Vacuum liquid chromatography (VLC): silica gel 60 H (Qingdao Marine Chemical Plant). TLC: precoated silica gel GF₂₅₄ (Qingdao Marine Chemical Plant), visualization by spraying with 1% vanillin in conc. H₂SO₄, followed by heating at 105° for 1–2 min. Optical rotation: Schmidt & Haensch Polartronic hnqw5 polarimeter. IR Spectra: Bruker EQUINOX-55 IR spectrophotometer, with KBr pellets; in cm⁻¹. NMR Spectra: Varian Unity INOVA-500 spectrometer operating at 500 (¹H) and 125 (¹³C) MHz, in CDCl₃; chemical

shifts δ in ppm rel. to Me₄Si (=0 ppm), coupling constants J in Hz. FAB-MS: VG~ZAB-HS and VG~Autospec-500~mass spectrometers; in m/z. GC/MS: Finnigan Voyager GC/MS apparatus (DB-5 MS column, 30 m \times 0.25 mm \times 0.25 mm \times 0.25 μ m). Elemental analysis: Elementar Vario EL CHNS elemental analyzer.

Animal Material. The marine sponge Ircinia fasciculata (PALLAS) was collected from Wei-zhou Island, Beihai, P. R. China, during September 2001, and authenticated by Mr. Zhi-Can Tang (Institute of Oceanology, Chinese Academy of Sciences). A voucher specimen (01-WZ-02) was deposited at the Research Center of Organic Natural Products, Sun Yat-Sen (Zhongshan) University.

Extraction and Isolation. The dried marine sponge (500 g) was extracted with 95% aq. EtOH at r.t. (3 \times). The EtOH solns, were evaporated in vacuo to afford 40.5 g of residue. The latter was suspended in H₂O, and successively extracted with AcOEt and BuOH. The AcOEt soluble fraction (28.6 g) was subjected to VLC using a step gradient of hexane/AcOEt. The fraction eluted with hexane/AcOEt 8:2 afforded 1 (5.4 mg), and the one eluted with hexane/AcOEt 6:4 yielded 2 (3.2 mg).

Ircisulfamide (= N-[(IS*,2S*,3R*)-2-Hydroxy-1-(hydroxymethyl)-3-(sulfooxy)heptadecyl]hexadecan-amide; **1**). Colorless, amorphous powder. [α] $_{0}^{20}$ = +16.7 (c = 0.24, CHCl $_{3}$). IR (KBr): 3383, 3312, 2919, 2850, 1645, 1545, 1470, 1382, 1234, 1082, 718. 1 H- and 13 C-NMR 1): see *Table 1*. FAB-MS: 636 ([M + H] $^{+}$). HR-FAB-MS: 636.4865 ([M + H] $^{+}$, C_{34} H $_{70}$ NO $_{7}$ S $^{+}$; calc. 636.4873).

Desulfation of Ircisulfamide (1). Compound 1 (4.0 mg) was dissolved in a mixture of anh. pyridine (0.5 ml) and 1,4-dioxane (0.5 ml), and heated at 120° for 18 h. The mixture was cooled to r.t. and then partitioned between CHCl₃ and H₂O. The org. layer afforded the desulfated product 1a (1.5 mg) as an amorphous powder. [α]_D²⁰ = +19.3 (c = 0.01, CHCl₃). ¹³C-NMR (CDCl₃)¹): 174.5 (C(1')); 76.5 (C(3)); 72.0 (C(4)); 62.1 (C(1)); 53.1 (C(2)); 36.3 (C(2')); 33.4 (C(5)); 29.7 – 29.4 (n CH₂); 25.8 (C(3')); 22.7 (C(17), C(15)); 14.1 (C(18), C(16')). FAB-MS: 557 ([M + H] $^+$).

Ircicerebroside (=(2R*)-N-[(1S*,2R*,3E,7E)-1-[(β-D-Glucopyranosyloxy)methyl]-2-hydroxy-8-methyl-heptadeca-3,7-dienyl]-2-hydroxyeicosanamide; **2**). Colorless, amorphous powder. [a] $_{0}^{20}$ = +9.1 (c = 0.02, MeOH). IR (KBr): 3378, 2921, 2851, 1643, 1536, 1456, 1080, 1036. 1 H- and 13 C-NMR 1): see *Table* 2. FAB-MS: 784 ([M+H] $^{+}$), 605 ([M+H - C $_{6}$ H $_{12}$ O $_{6}$] $^{+}$). HR-FAB-MS: 784.6280 ([M+H] $^{+}$, C $_{45}$ H $_{86}$ NO $_{9}^{+}$; calc. 784.6303)

Methanolysis. Compounds 1 or 2 (1.1 mg) in a mixture of 1N aq. HCl (5 ml) and MeOH (15 ml) was heated at reflux for 15 h with magnetic stirring. Then, H_2O (25 ml) was added, and the mixture was extracted with hexane (3×). The combined org. layer was concentrated under N_2 to yield a colorless solid, which was identified by GC/MS analysis. In the case of 2, the aq. layer was removed under N_2 , and purified on a C18 reversed-phase column to afford methyl glucopyranoside (0.2 mg). $[\alpha]_{D}^{20} = +76.2$ (c = 0.02, MeOH).

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